Densities and magnetization for metals and insulators. PBC and the simulation of systems with reduced periodicity

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Outline

- Densities
- Magnetic vs non magnetic
- Metal vs Insulators
- Potentials definition and meaning
- Non periodic systems with PBC formalism



Densities

- Non magnetic system: total density $n(\mathbf{r})$
- Magnetic system with collinear spin:

$$n({f r})=n_{\uparrow}({f r})+n_{\downarrow}({f r}) \qquad m({f r})=n_{\uparrow}({f r})-n_{\downarrow}({f r})$$

• Magnetic noncollinear spin:

$$n(\mathbf{r}) \qquad \quad \{m_x(\mathbf{r}), m_y(\mathbf{r}), m_z(\mathbf{r})\}$$



PW cutoff for densities

• The cutoff lengt for densities is by construction 2 the one for wavefunctions. It means cutoff energy is 4 times.

 $n_v(\mathbf{r}) = \sum_{n,\mathbf{k}} \phi^*_{n,\mathbf{k}} \phi_{n,\mathbf{k}}$

• In case of USPP or PAW we have also augmentation charges and ecutrho is tipically about 6 - 10 times larger than ecut.

$$n_v(\mathbf{r}) = \sum_{n,\mathbf{k}} \phi^*_{n,\mathbf{k}} \phi_{n,\mathbf{k}} + \sum_{ij}
ho_{ij} Q_{ij}(\mathbf{r}) \qquad
ho_{ij} = \sum_{n,\mathbf{k}} \langle \phi_{n,\mathbf{k}} | eta_i
angle \langle eta_j | \phi_{n,\mathbf{k}}
angle$$



Non magnetic (nspin=1)

- The SCF loop starts from a non magnetic configuration
- Each Kohn-Sham state has an occupation from 0 to 2

$$n(r) = 2 * \sum_{n, \mathbf{k}} f_{n, \mathbf{k}} \phi^*_{n, \mathbf{k}} \phi_{n, \mathbf{k}}$$

• Non magnetic systems may be also studied with the noncollinear formalism (spinors). In this case each KS spinor state is doubly degenerate.



Collinear magnetic (nspin = 2)

- distinct KS states are used 2 different polarizations(spin up, spin down)
- The magnetization is a scalar

$$n(r)^{\uparrow} = \sum_{n, \mathbf{k}} f_{n, \mathbf{k}, \uparrow} \phi^*_{n, \mathbf{k}, \uparrow} \phi_{n, \mathbf{k}, \uparrow} \quad n(r)^{\downarrow} = \sum_{n, \mathbf{k}} f_{n, \mathbf{k}, \downarrow} \phi^*_{n, \mathbf{k}, \downarrow} \phi_{n, \mathbf{k}, \downarrow}$$

- One needs to specify a starting magnetization or fix a total magnetization for the system.
 - Starting magnetization: the scf is free to converge to any magnetization
 - Total magnetization, occupations for majority and minority spin are computed separately in order to keep to specified total magnetization during the scf loop



Non collinear magnetism

• Magnetization is a vector.

$$ec{\mathbf{m}}(r) = \sum_{n,\mathbf{k}} f_{n,\mathbf{k}} \phi^*_{n,\mathbf{k}} ec{\sigma} \phi_{n,\mathbf{k}}$$

• It is also possible to introduce the spin-orbit coupling. One has to use fully relativistic pseudo-potentials.



Metals vs insulators

- Insulators: there is a finite gap between occupied and emply levels it is possile to determine a number for occupied levels fixed for all k-points.
- Metals: there is no gap between occupied and empty levels, to make feasible the k-point summation is necessary use the smearing technique;
- The smearing at the Fermi level increases the total energy, this increase should be kept small.



from: M.Methfessel, A.T. Paxton PRB 40, 3616 (1989)

Occupations for magnetic systems

- One has to smearing when the scf loop is free to search for the ground state magnetization
- It is possible to use fixed occupations if total_magnetization is specified with an integer value.
- for atoms it is possible to specify occupations in input directly

Potentials

- V_{Hartree}: The electrostatic potential generated by the electronic density. Long range
- V_{xc} : The exchange correlation potential, short range.
- V_{loc}: The long range part of the ion-electron interaction.
- E_{Ewald}: the electrostatic energy of the ion-ion interaction.

Electrostatics in PBC

- The net charge per unit cell must be 0 otherwise the energy would diverge.
- All potentials must be periodic, the net value of the electric field must be vanishing
- There is no direct reference the the vacuum level.
- The electrostatic moments per unit cell are generally ill defined.

Picture from: Electronic Structure: Basic Theory and Practical Methods R.M. Martin Cambridge University Press (2008)

Picture from: G. Makov, M.C. Payne PRB 51, 4014 (1995)

Electrostatics in PBC: Hartree potential

• The Hartree potential is computed as:

$$v_{H}(\mathbf{r}) = 4\pi e^{2}\sum_{\mathbf{G}
eq 0}rac{n(\mathbf{G})}{\left|\mathbf{G}
ight|^{2}}e^{i\mathbf{G}\cdot\mathbf{r}}$$

• It is like we add a constant background $\,n_+$ such that

$$n'(G\equiv 0)=\int_{cell}{(n({f r})+n_+)d^3r=0}$$

Ionic electrostatic energy: (Ewald sum)

- The ionic electrostatic energy is computed as the energy of point positive changes immersed in a uniform negative compensating background (Ewald sum)
- If the system is neutral the two compensating backgrounds cancel each other
- If the system is charged it will be like we had added a non vanishing compensating background.

Picture from: Electronic Structure: Basic Theory and Practical Methods R.M. Martin Cambridge University Press (2008)

Ion-electron interaction

• The ion electron interaction is given by the pseudopotential

$${\hat V}_{pseudo} = V_{loc}(r) + \sum \ket{eta_i} D_{ij} ig \langle eta_i
vert$$

- The non local part vanishes beyond the cutoff radii of the pseudopotential.
- The local part contains the long range coulombian part of the lon electron interaction

Potentials Si:

Simulating systems with reduced periodicity.

- Molecules and clusters. Finite systems with vanishing systems with the periodic images
- 1D systems: Periodicity along 1 direction: nanotube, nanorods etc/
- 2D systems: Periodicity along 2 dimensions with a vacuum layer between slabs.
- Mixed systems: structures embedded in a periodic system, point defects, interfaces, molecules on surfaces.

Makov Payne corrections

- Neutral system
 - No dipole the energy c or cubic lattice onvergese as $\ {\cal O}(L^{-5})$ dipole in non cubic system $\ {\cal O}(L^{-3})$ Ο
 - 0
- Charged system

0

$$E = E_0 - rac{q^2 lpha}{2L} - rac{2 \pi q Q}{3L^3} + \mathcal{O}(L^{-5})$$

Polar surfaces

 In Polar 2D systems we can eliminate the artificial electric field with a sawtooth potential.

Dipole correction

Water molecule

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That's all !!! Thanks for you attention

